GUJARAT TECHNOLOGICAL UNIVERSITY Pharm.D. 1st year

Subject Name: Medicinal Biochemistry Subject Code: 818803

Tea	aching Sch	eme (Hours	s)	Evaluation Scheme ()			ks)	Total Marks
Theory	Tutorial	Practical	Total	Theory		Practical		
				External	Internal	External	Internal	
3	1	3	7	70	30	70	30	200

Sr. No	Course Contents	Hours	Module Weightage
1	Introduction to biochemistry: Cell and its biochemical organization, transport process across the cell membranes. Energy rich compounds; ATP, Cyclic AMP and their biological significance.	5	5.5%
2	Enzymes : Definition; Nomenclature; IUB classification; Factor affecting enzyme activity; Enzyme action; enzyme inhibition. Isoenzymes and their therapeutic and diagnostic applications; Coenzymes and their biochemical role and deficiency diseases.	7	8%
3	Carbohydrate metabolism : Glycolysis, Citric acid cycle (TCA cycle), HMP shunt, Glycogenolysis, gluconeogenesis, glycogenesis. Metabolic disorders of carbohydrate metabolism (diabetes mellitus and glycogen storage diseases); Glucose, Galactose tolerance test and their significance; hormonal regulation of carbohydrate metabolism.	16	18%
4	Lipid metabolism: Oxidation of saturated (\Box -oxidation); Ketogenesis and ketolysis; biosynthesis of fatty acids, lipids; metabolism of cholesterol; Hormonal regulation of lipid metabolism. Defective metabolism of lipids (Atheroslerosis, fatty liver, hypercholesterolmiea).	12	13%
5	Biological oxidation: Coenzyme system involved in Biological oxidation. Electron transport chain (its mechanism in energy capture; regulation and inhibition); Uncouplers of ETC; Oxidative phosphorylation;	7	8%
6	Protein and amino acid metabolism: protein turn over; nitrogen balance; Catabolism of Amino acids (Transamination, deamination & decarboxylation). Urea cycle and its metabolic disorders; production of bile pigments; hyperbilirubinemia, porphoria, jaundice. Metabolic disorder of Amino acids.	12	13%
7	Nucleic acid metabolism: Metabolism of purine and pyrimidine nucleotides; Protein synthesis; Genetic code; inhibition of protein synthesis; mutation and repair mechanism; DNA replication (semiconservative /onion peel models) and DNA repair mechanism.	10	11%
8	Introduction to clinical chemistry: Cell ; composition; malfunction; Roll of the clinical chemistry laboratory.	2	2%

9	The kidney function tests: Role of kidney; Laboratory tests for	5	5.5%
	normal function includes- a) Urine analysis (macroscopic and		
	physical examination, quantitative and semiquantitative tests.) b)		
	Test for NPN constituents. (Creatinine /urea clearance,		
	determination of blood and urine creatinine, urea and uric acid) c)		
	Urine concentration test d) Urinary tract calculi. (stones)		
10	Liver function tests: Physiological role of liver, metabolic,	5	5.5%
	storage, excretory, protective, circulatory functions and function in		
	blood coagulation. a) Test for hepatic dysfunction-Bile pigments		
	metabolism. b) Test for hepatic function test- Serum bilirubin,		
	urine bilirubin, and urine urobilinogen. c) Dye tests of excretory		
	function. d) Tests based upon abnormalities of serum proteins.		
	Selected enzyme tests.		
11	Lipid profile tests: Lipoproteins, composition, functions.	3	3.5%
	Determination of serum lipids, total cholesterol, HDL cholesterol,		
	LDL cholesterol and triglycerides.		
12	Immunochemical techniques for determination of hormone levels	3	3.5%
	and protein levels in serum for endocrine diseases and infectious		
	diseases		
	Radio immuno assay (RIA) and Enzyme Linked Immuno Sorbent		
	Assay (ELISA)		
13	Electrolytes: Body water, compartments, water balance, and	3	3.5%
	electrolyte distrubution. Determination of sodium, calcium		
	potassium, chlorides, bicarbonates in the body fluids.		

Course Materials:

Text books (Theory)

- a. Harpers review of biochemistry Martin
- b. Text book of biochemistry U.Satyanarayana
- c. Text book of clinical chemistry- Alex kaplan&LaverveL.Szabo

Reference books (Theory)

- a. Principles of biochemistry –Lehninger
- b. Text book of biochemistry –Ramarao

Pharm.D. 1st year MEDICINAL BIOCHEMISTRY

Practical (3 Hours/ Week, 6 Credits, 90 Hours)

Sr. No.	Experiments
1	Qualitative analysis of normal constituents of urine.*
2	Qualitative analysis of abnormal constituents of urine.*
3	Quantitative estimation of urine sugar by Benedict's reagent method.**
4	Quantitative estimation of urine chlorides by Volhard's method.**

5	Quantitative estimation of urine creatinine by Jaffe's method.**
6	Quantitative estimation of urine calcium by precipitation method.**
7	Quantitative estimation of serum cholesterol by LibermannBurchard's method.**
8	Preparation of Folin Wu filtrate from blood.*
9	Quantitative estimation of blood creatinine.**
10	Quantitative estimation of blood sugar Folin-Wu tube method.**
11	Estimation of SGOT in serum.**
12	Estimation of SGPT in serum.**
13	Estimation of Urea in Serum.**
14	Estimation of Proteins in Serum.**
15	Determination of serum bilirubin**
16	Determination of Glucose by means of Glucoseoxidase.**
17	Enzymatic hydrolysis of Glycogen/Starch by Amylases.**
18	Study of factors affecting Enzyme activity. (pH& Temp.)**
19	Preparation of standard buffer solutions and its pH measurements (any two)*
20	Experiment on lipid profile tests**
21	Determination of sodium, calcium and potassium in serum.**

** indicate major experiments & * indicate minor experiments

Scheme of Practical Examination

	Internal/ Sessional	External
Synopsis	05	15
Major Experiment	10	25
Minor Experiment	03	15
Viva	02	15
Max. marks	20	70
Duration	3 hours	4 hours

Note: Total sessional marks is 30 (20 for practical sessional plus 10 marks for regularity, promptness, viva-voce and record maintenance)

Course Materials:

Reference books (Practical)

- a. Practical Biochemistry-David T.Plummer.
- b. Practical Biochemistry-Pattabhiraman.